

# Some Green Analytical Approaches

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2005-2023

# Sample pre-treatment

## ✓ **No sample pre-treatment**

- Matrix effect
- Low concentration of the analyte

## ✓ **The main aim**

- Clean up
- Pre-concentration of the analytes of interest
- Compatible with the analytical system

## ✓ **Choice of the technique**

- Nature of analyte
- Nature of sample
- Detection technique used

# Liquid–liquid extraction (LLE)

**One of the oldest separation technique**

## ✓ **Advantages**

- a simple technique
- **wide range of available organic solvents**

## ✓ **Disadvantages**

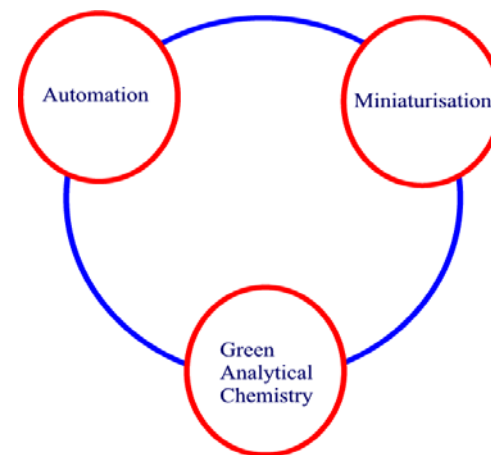
- time-consuming
- tedious
- **large volumes of organic solvents used**
- **production of vast amounts of organic waste**

# Improvement of conventional LLE

✓ **Miniaturization**

✓ **Automation**

✓ **Convenient to green analytical chemistry**



# Liquid-phase microextraction

## ✓ **LPME**

- **Small amount of solvent used**
- **High sample-to-acceptor volume ratio**

## ✓ **Main categories:**

- **SDME**, Single Drop Microextraction
- **HF-LPME**, Hollow-Fiber Liquid Phase Microextraction
- **DLLME**, Dispersive Liquid–Liquid Microextraction
- **HLLME**, Homogeneous Liquid–Liquid (micro)Extraction

# What is the microextraction?

„... solvent microextraction (SME) is a technique of sample preparation by extraction ... with solvent volumes of 100  $\mu\text{L}$  or less.“



# Dispersive liquid-liquid microextraction

DLLME

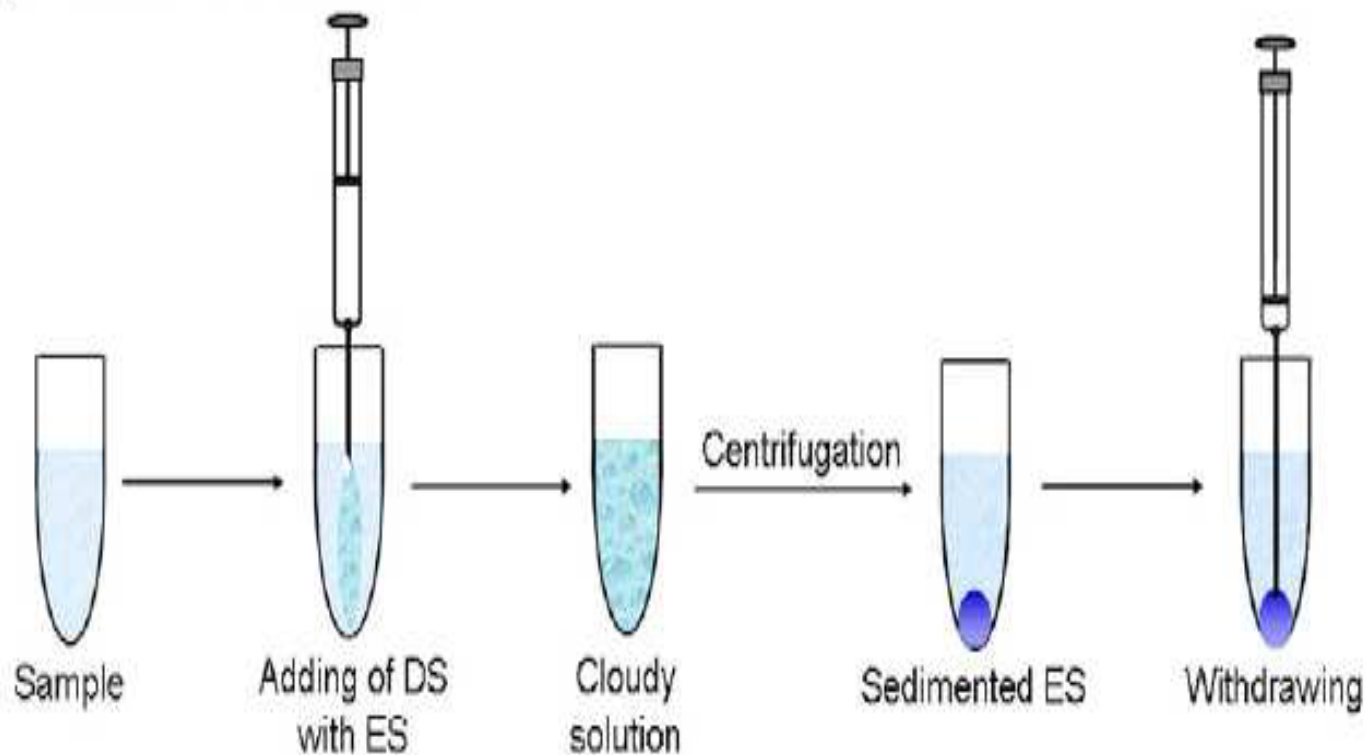
# DLLME

- ✓ Developed by Rezaee et al. in 2006
- ✓ A ternary component system
  - sample solution
  - extraction solvent – **density higher than that of water**
  - dispersive solvent – **miscible with both water and extraction solvent**
- ✓ When injected into the sample, a cloudy solution is formed and a large surface area between extraction solvent and aqueous sample enables quickly achieving of equilibrium

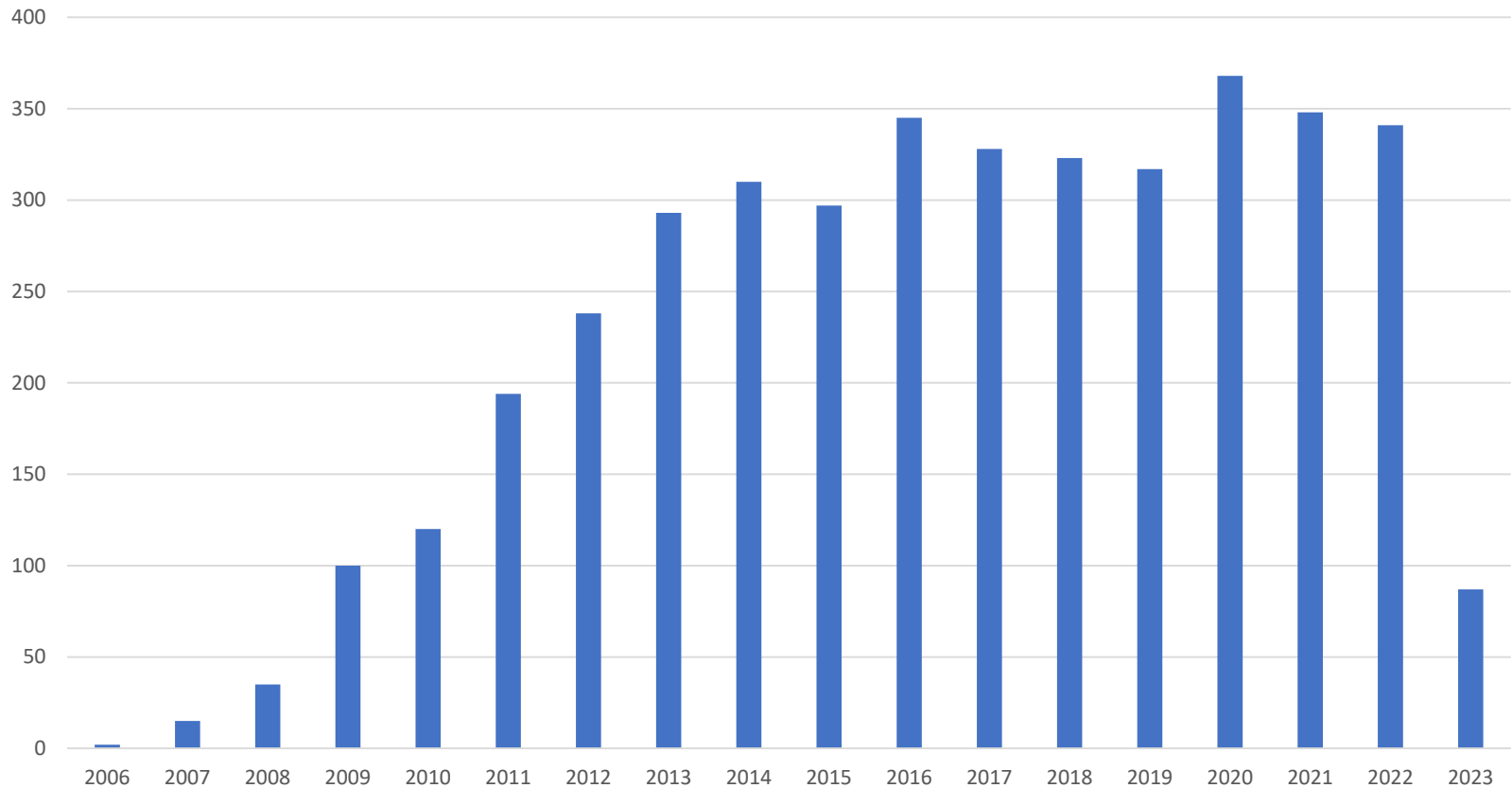




# DLLME – schematic



# Evolution of number of papers



# Factors affecting DLLME

- ✓ pH
- ✓ Ionic strength
- ✓ Extraction solvent nature and volume
- ✓ Dispersive solvent nature and volume
- ✓ Auxiliary agents
- ✓ Auxiliary energies
- ✓ Extraction time

# DLLME

## ✓ Advantages

- quick achieving of the equilibrium
- very short extraction time
- simplicity and low costs

## ✓ Disadvantages

- limitations to solvents with *density higher than water*
- problems with *dispersive solvents*
- time-consuming *centrifugation* step

# DLLME

- ✓ Therefore there were new techniques developed
  - allow the use of extraction *solvents lighter than water*
  - allow to *omit the dispersive solvent*
  - allow to *omit centrifugation step*

# Extraction solvent

## ✓ **Requirements for extraction solvents**

- density higher than that of water
  - ✓ chlorobenzene, chloroform, tetrachloromethane, tetrachloroethane

## ✓ **Advantages** of using an extraction solvent heavier than water

- easy removal of the extraction phase after extraction

## ✓ **Disadvantages** of using an extraction solvent heavier than water

- hazardous

## ✓ **Solution**

- use of *extraction solvents lighter than water*

# DLLME

- ✓ The use of extraction ***solvents lighter than water***
  - use of *special extraction vessels*
    - ✓ **home-made device is needed**
    - ✓ **tedious**
    - ✓ **laborious**
  - use of „solidification of floating organic droplet“ (*DLLME-SFO*)
    - ✓ **limited number of solvents with melting point between 10-30 °C**
  - use of *auxiliary solvent* for adjustment of solvents mixture density

# Dispersive solvent

- ✓ **Requirements** for dispersion solvents
  - miscibility with both the sample and the extraction solvent
    - ✓ methanol, ethanol, acetone, acetonitrile
  
- ✓ **Advantages** of using a dispersion solvent
  - increasing the extraction efficiency (*formation of a cloudy state*)
  
- ✓ **Disadvantages** of using a dispersion solvent
  - reducing the extraction efficiency of *polar analytes*
  
- ✓ **Solution**
  - *DLLME without the use of dispersive solvent*



# Without dispersive solvent

## ✓ *Alternatives to the use of dispersive solvent*

- ultrasonication
- vortex mixing
- air-assisted DLLME
- adding of surfactants
- magnetic stirring

# Centrifugation

- ✓ **Requirements** for centrifugation
  
- ✓ **Advantages** of using centrifugation
  - perfect separation of the aqueous and organic phases
  
- ✓ **Disadvantages** of using centrifugation
  - time-consuming step (**2-20 min**)
  
- ✓ **Solution**
  - *DLLME without using centrifugation*

# DLLME without centrifugation

- ✓ Solvent terminated DLLME

# DLLME vs DLPME

- ✓ We should distinguish between **two techniques**:
  - 1) dispersive liquid-liquid microextraction (**DLLME**), and
  - 2) dispersive liquid-phase microextraction (**DLPME**).

# DLLME vs DLPME

## ✓ DLLME

- use a *mixture of extraction and dispersive solvents* and which in some instances, in addition to these two solvents, the formation of the cloudy solution is enhanced by adding *supplementary reagents* or applying *supplementary energy*

## ✓ DLPME

- *no dispersive solvent* is used

# Acronyms

- ✓ There is a tendency when researchers lightly modify an existing general method to give it
  - a **new name** and
  - a **new acronym**,
- ✓ which greatly complicates a search in the literature.



# Acronyms

- ✓ Ideally, the method name and its acronym should provide the reader **sufficient information**;
- ✓ however, it should also be **as simple as possible** and not burdened with unnecessary details.
- ✓ probable **no ideal state is possible**.



# Single drop microextraction

## ✓ Modalities

- **DI-SDME**, Direct Immersion Single Drop Microextraction
- **HS-SDME**, Headspace Single Drop Microextraction
- **LLLME**, Liquid-liquid-liquid Microextraction



# Single drop microextraction

## ✓ **Disadvantages**

- Drop instability
- Time-consuming
- Low volume of the sedimented phase
- Viscosity of the sedimented phase

## ✓ **Solution**

- evaporation and reconstitution of the sedimented phase
- back extraction step
- dilution of the sedimented phase

# Using an optical probe as the microdrop holder in headspace single drop microextraction

Determination of sulfite in food samples

S. Zaruba, A.B. Vishnikin, J. Škrlíková, V. Andruch, *Using an optical probe as the microdrop holder in headspace single drop microextraction: Determination of sulfite in food samples*, Anal. Chem., 88 (2016) 10296-10300



# HS-SDME

- ✓ A novel headspace single-drop microextraction method (HS-SDME) for determination of sulfite was developed.
- ✓ An optical probe was used as the droplet holder in the HS-SDME procedure, and the analytical signal (absorbance) was monitored online during the extraction process.

# A two-in-one device for online monitoring of direct immersion single-drop microextraction

An optical probe as both microdrop holder and measuring cell

S. Zaruba, A.B. Vishnikin, J. Škrlíková, A. Diuzheva, I. Ozimaničová, K. Gavazov, V. Andruch, *A two-in-one device for online monitoring of direct immersion single-drop microextraction: An optical probe as both microdrop holder and measuring cell*, RSC Adv., 7 (2017) 29421-29427



# DI-SDME

- ✓ An optical probe is proposed as the microdrop holder and simultaneously the measuring cell in a direct immersion single-drop microextraction (DI-SDME) procedure.
- ✓ This approach enables the analytical signal (absorbance of organic phase) to be monitored online during the extraction process.
- ✓ Based on the suggested approach, a novel DI-SDME method for the determination of thiocyanate ions in human saliva samples.

# Optical probe – advantages

- ✓ optical probe as the holder of the extraction drop in SDME
- ✓ optical probe as the measuring cell in SDME
- ✓ necessity of transferring the extraction phase to a microcuvette is eliminated
- ✓ allows the absorbance to be recorded online
- ✓ allows the stirring rate to be increased
- ✓ setup does not include any homemade components

# Green solvents

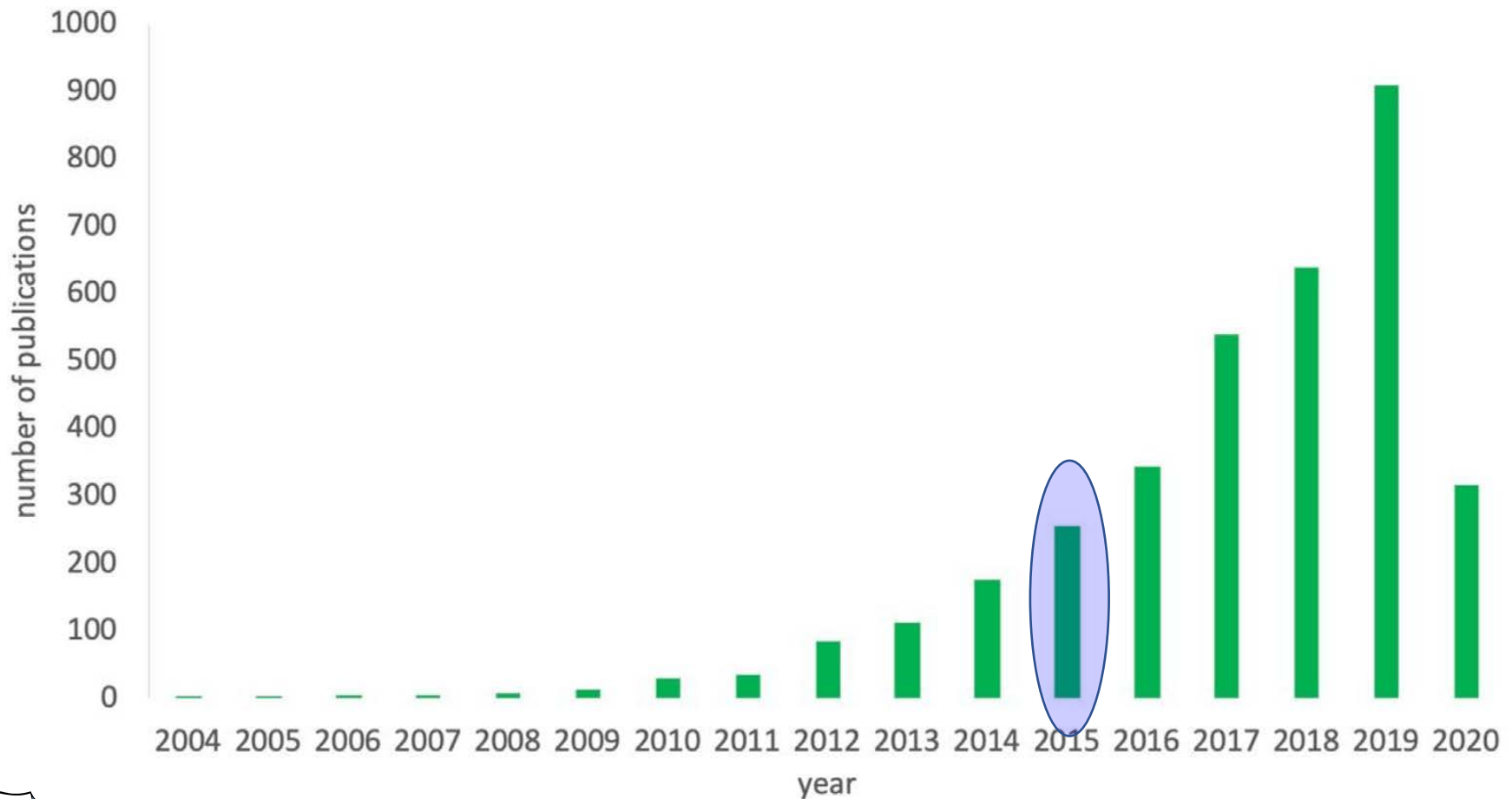
- ✓ Ionic liquids (IL)
- ✓ **Deep eutectic solvents (DES)**
- ✓ Switchable-hydrophilicity solvents (SHS)
- ✓ Surfactants

# Deep eutectic solvents

- ✓ DES is a mixture of two or more pure compounds for which the eutectic point temperature is lower than that of the ideal liquid mixture, representing a significant negative deviation from ideality.



# Deep eutectic solvents



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